

Diastereoselective Synthesis of $\psi[(E)\text{-CH=CMe}]$ - and $\psi[(Z)\text{-CH=CMe}]$ -Type Dipeptide Isosteres by Organocopper-Mediated *anti*- S_N2' Reaction

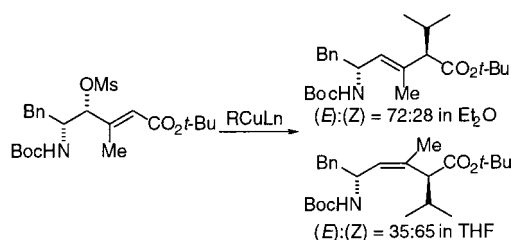
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ABSTRACT



Acyclic $\psi[(E)\text{-CH=CMe}]$ - and $\psi[(Z)\text{-CH=CMe}]$ -type dipeptide isosteres were efficiently synthesized. In a key reaction, α -alkylation of γ -mesyloxy- β -methyl- α,β -unsaturated esters with organocyanocuprates in diethyl ether or tetrahydrofuran preferentially afforded the $\psi[(E)\text{-CH=CMe}]$ - or $\psi[(Z)\text{-CH=CMe}]$ -isomer, respectively, via *anti*- S_N2' mechanism.

Bioisosteres of natural amino acids and dipeptides are useful tools for investigation of molecular recognition including receptor/ligand and enzyme/substrate interactions, and they represent diverse elements of practical value for constructing chemical libraries in medicinal chemistry.¹ (*E*)-Alkene dipeptide isosteres (EADIs) are examples of structures having modified peptide backbones, which are designed to provide resistance to biodegradation² and conformational restriction of a β -turn substructures,³ etc. We and others have reported the stereoselective synthesis of EADIs mediated by organocopper reagents⁴ and their application to bioactive peptides.^{2,5} For instance, we recently synthesized an EADI-containing

cyclic RGD pseudopeptide **3** and evaluated its biological activities.⁶ A D-Phe- $\psi[(E)\text{-CH=CH}]$ -L-Val isostere was

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incorporated as an equivalent to the $(i + 1)-(i + 2)$ position in the type II' β -turn substructure of the cyclic RGD pentapeptide **1** reported by Kessler et al. (Figure 1).⁷ In the

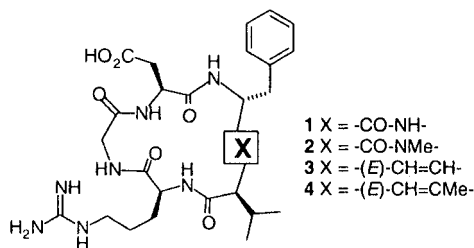
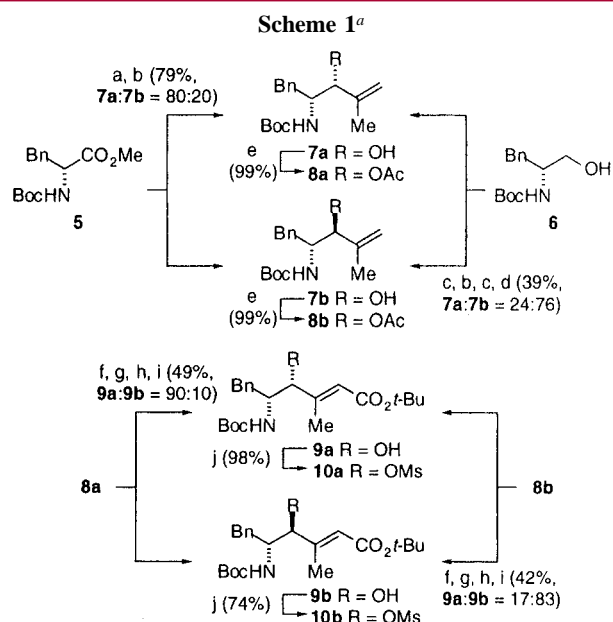


Figure 1. Cyclic RGD peptides **1** and **2** by Kessler et al. and EADI-containing cyclic RGD pseudopeptides **3** and **4**.

inhibition assay of vitronectin- $\alpha_v\beta_3$ integrin binding, the potency of pseudopeptide **3** was higher than that of **1** and nearly equal to that of the improved cyclic peptide **2**.⁸

While the reason for the high activity of the pseudopeptide **3** has not been elucidated yet, incorporation of the D-Phe- $\psi[(E)\text{-CH=CH}]$ -L-Val moiety may convert the peptide backbone to a more active form, as well as *N*-methylvaline. Thus, we designed the pseudopeptide **4** containing the D-Phe- $\psi[(E)\text{-CH=CMe}]$ -L-Val moiety in order to rationally investigate the effect of both the *N*-methylvaline and the (*E*)-alkene moiety. Synthesis of **4** required preparation of this unique dipeptide isostere. In this communication, we describe the first unequivocal synthesis of a $\psi[(E)\text{-CH=CMe}]$ -type dipeptide isostere and its $\psi[(Z)\text{-CH=CMe}]$ -congener, which is inherently an unanticipated product, obtained with organocopper reagents.

Synthesis of $\psi[(E)\text{-CH=CMe}]$ -Type Dipeptide Isostere Precursors. Synthesis of key intermediate, γ -mesyloxy- α,β -unsaturated esters **10**, started from D-phenylalanine derivative **5** or D-phenylalaninol derivative **6**. This provided a generalized synthetic strategy toward $\psi[(E)\text{-CH=CMe}]$ -type dipeptide isosteres from chiral amino acids (Scheme 1). A *syn*-allyl alcohol **7a** was stereoselectively prepared by reduction of Boc-D-Phe-OMe **5** with DIBAL-H at -78°C in CH_2Cl_2 /toluene, followed by treatment with isopropenyl Grignard



^a (a) DIBAL-H, CH_2Cl_2 /toluene. (b) $\text{CH}_2=\text{CMeMgCl}\cdot\text{ZnCl}_2\cdot\text{LiCl}$, THF. (c) $(\text{COCl})_2$, DMSO, DIEA, CH_2Cl_2 . (d) $\text{Zn}(\text{BH}_4)_2$, Et_2O . (e) Ac_2O , pyridine, DMAP, CHCl_3 . (f) O_3 , EtOAc . (g) DMS. (h) $\text{Ph}_3\text{P}=\text{CHCO}_2t\text{-Bu}$, CHCl_3 , reflux. (i) Na_2CO_3 , MeOH. (j) MsCl, TEA, THF.

reagent (*syn:anti* = 80:20). Alternatively, an *anti*-allyl alcohol **7b** was preferentially afforded by reduction of the enone (obtained by Swern oxidation of a diastereomixture of allyl alcohols **7a** and **7b**) with $\text{Zn}(\text{BH}_4)_2$ in Et_2O (*syn:anti* = 24:76).⁹ The diastereomerically pure alcohols **7a** and **7b** could be separated by flash chromatography over silica gel followed by recrystallization, respectively.

After protection of hydroxyl groups, acetates **8a** and **8b** were converted to α,β -unsaturated esters. Ozonolysis of **8a** followed by reductive treatment with dimethyl sulfide and successive Wittig reaction with $\text{Ph}_3\text{P}=\text{CHCO}_2t\text{-Bu}$ gave γ -acetoxy- α,β -unsaturated esters. Deprotection of acetyl groups yielded two isomers of the γ -hydroxy- α,β -unsaturated esters. Unexpectedly, the resulting minor isomer was not the (*Z*)-isomer of *syn*- α,β -unsaturated esters **9a** but rather the *anti*-(*E*)-isomer **9b**, which apparently originated as a result of epimerization at the chiral center of the acetoxy group (**9a:9b** = 90:10). Even in the case of the acetate **8b**, both isomers of α,β -unsaturated esters **9a** and **9b** were obtained in similar manner (**9a:9b** = 17:83). Each isomer of esters **9a** and **9b** was readily purified by flash chromatography. The *E* geometry and the relative configuration of the hydroxy groups of α,β -unsaturated esters **9a** and **9b** were established by ^1H NMR analysis of the corresponding acetanilides.¹⁰ Esters **9a** and **9b** were converted into the respective mesylates **10a** and **10b**.

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Table 1. Alkylation of *syn*- γ -Mesyloxy- β -methyl- α,β -unsaturated Ester **10a** with Organocyanocuprates

entry	reagent ^a	additive ^b	solvent	condition	product ratio ^c	yield ^e (%)
					11a:11b:11c+11d^d:12	
1	<i>i</i> -PrCu(CN)MgCl·BF ₃		THF	0 °C, 0.5 h	40:51: - <i>g</i> :9	82
2	<i>i</i> -Pr ₂ Cu(CN)(MgCl) ₂ ·BF ₃		THF	-78 °C, 0.5 h	35:64: - <i>g</i> :1	94
3	<i>i</i> -Pr ₂ Cu(CN)(MgCl) ₂ ·BF ₃	HMPA	THF	-78 °C, 0.5 h	28:71: - <i>g</i> :1	91
4	<i>i</i> -Pr ₂ Cu(CN)(MgCl) ₂ ·BF ₃	TMEDA	THF	-78 °C, 0.5 h	27:72: - <i>g</i> :1	92
5	<i>i</i> -Pr ₂ Cu(CN)(MgCl) ₂	18-crown-6	THF	-78 °C, 0.5 h	25:73: - <i>g</i> :1	86
6	<i>i</i> -PrCu(CN)MgCl·BF ₃		Et ₂ O	0 °C, 3 h	36:63: - <i>g</i> :1	15 ^h
7	<i>i</i> -Pr ₂ Cu(CN)(MgCl) ₂ ·BF ₃		Et ₂ O	-78 °C, 0.5 h	46:27:26:1	90
8	<i>i</i> -Pr ₂ Cu(CN)(MgCl) ₂ ·BF ₃	HMPA	Et ₂ O	-78 °C, 0.5 h, then 0 °C, 0.5 h	70:27: - <i>g</i> :3	93
9	<i>i</i> -Pr ₂ Cu(CN)(MgCl) ₂ ·BF ₃	TMEDA	Et ₂ O	-78 °C, 0.5 h	44:25:28:3	81
10	<i>i</i> -Pr ₂ Cu(CN)(MgCl) ₂	18-crown-6	Et ₂ O	-78 °C, 0.5 h	14:84: - <i>g</i> :2	93
11	<i>i</i> -Pr ₂ Cu(CN)(MgCl) ₂ ·BF ₃	THF	Et ₂ O	-78 °C, 0.5 h	43:42:12:3	97
12	<i>i</i> -Pr ₂ Cu(CN)(MgCl) ₂ ·BF ₃		Et ₂ O/THF ^f	-78 °C, 0.5 h	37:62: - <i>g</i> :1	87

^a All reactions were carried out with 4 molar equiv of reagent. ^b 4 molar equiv. ^c Product ratios were determined by HPLC and ¹H NMR. ^d Containing a small amount of an unknown product. ^e Combined yield. ^f Et₂O/THF = 7:3. ^g Although we cannot conclusively rule out its presence, we failed to isolate the corresponding product. ^h The starting material was recovered (77%).

Synthesis of $\psi[(E)\text{-CH=CMe}]$ - and $\psi[(Z)\text{-CH=CMe}]$ -Type Dipeptide Isosteres. Alkylation of γ -Mesyloxy- β -methyl- α,β -unsaturated Esters with Organocyanocuprates. In the synthesis of chiral α -alkyl- β,γ -unsaturated esters^{11a} and chiral α,α -dialkyl- β,γ -unsaturated esters^{11b} and its application to the synthesis of $\psi[(E)\text{-CH=CH}]$ -type dipeptide isosteres,^{4a} the regio- and stereochemical outcomes of organocopper-mediated alkylation of γ -mesyloxy- α,β -unsaturated esters are fully documented. Alkylation proceeds through an *anti*-S_N2' mechanism, and all products are exclusively of *E*-geometry. As such, it was our expectation that $\psi[(E)\text{-CH=CMe}]$ -type dipeptide isosteres would be easily prepared from γ -mesyloxy- β -methyl- α,β -unsaturated esters such as **10a** and **10b**.

However, treatment of **10a** with a "lower-order" organocyanocuprate-BF₃ complex, *i*-PrCu(CN)MgCl·BF₃, in THF, which is the usual condition for preparation of $\psi[(E)\text{-CH=CH}]$ -type dipeptide isosteres, gave D-Phe- $\psi[(Z)\text{-CH=CMe}]$ -L-Val-type dipeptide isostere **11b** as a major product, along with the expected D-Phe- $\psi[(E)\text{-CH=CMe}]$ -L-Val-type dipeptide isostere **11a** (**11a:11b** = 44:56, Table 1, entry 1).¹² A "higher-order" cyanocuprate-BF₃ complex, *i*-Pr₂Cu(CN)(MgCl)₂·BF₃ (which gave only reductive products in the case of alkylation of β -aziridinyl- α,β -unsaturated esters^{4d}), also yielded **11b** with higher *Z*-selectivity (**11a:11b** = 35:65, entry 2). Several additives (HMPA, TMEDA, 18-crown-6), which were intended to improve *E*-selectivity, further increased *Z*-selectivity (entries 3–5).

Next, we investigated conditions using only Et₂O as solvent. We originally thought this to be unsuitable for the alkylation because of sluggishness.^{4,11} Treatment with *i*-PrCu(CN)MgCl·BF₃ afforded a mixture of S_N2'-alkylated products in low yield (15%), and the substrate **10a** was recovered (77%, entry 6). The more reactive "higher-order" reagent *i*-Pr₂Cu(CN)(MgCl)₂·BF₃ in Et₂O gave alkylated products in excellent yield (90%) with concomitant formation of *anti*- and *syn*-S_N2' products^{4b} (*anti*-S_N2':*syn*-S_N2' = 74:26), and the *E*-ratio of the isomers was remarkably increased (**11a:11b** = 63:37) compared with the reaction in THF (entry 7). In addition, various additives were examined (entries 8–10), including HMPA, which proved to suppress formation of *syn*-S_N2' products, **11c** and **11d**, and apparently improved the overall yield of the D-Phe- $\psi[(E)\text{-CH=CMe}]$ -L-Val-type dipeptide isostere **11a** (**11a:11b** = 72:28, entry 8). In contrast, addition of 18-crown-6 decreased the formation of the *E*-isomer **11a** (entry 10).

To evaluate solvent effects on selectivity, alkylation in mixed solvents of Et₂O and THF was examined (entries 11 and 12). Addition of 4 equiv of THF in Et₂O decreased *E*-selectivity (**11a:11b** = 51:49), and a 7:3 mixture of Et₂O and THF yielded a ratio of **11a** and **11b** products similar to that using THF alone (**11a:11b** = 37:63). Taken together, cyclic ethers such as THF and 18-crown-6 probably affected the *Z*-selectivity in alkylation for reasons that are unclear.

Alkylation of the *anti*-isomer **10b** with organocyanocuprates was also investigated similarly (Table 2). In THF, both *i*-PrCu(CN)MgCl·BF₃ and *i*-Pr₂Cu(CN)(MgCl)₂·BF₃ provided **11c** and **11d** *Z*-selectively (entries 1 and 2). Whereas treatment of **10b** with *i*-PrCu(CN)MgCl·BF₃ in Et₂O resulted in recovered substrate with no conversion to products (entry 3), *i*-Pr₂Cu(CN)(MgCl)₂·BF₃ in Et₂O gave a D-Phe- $\psi[(Z)\text{-CH=CMe}]$ -L-Val-type dipeptide isostere **11d** as a main

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Table 2. Alkylation of *anti*- γ -Mesyloxy- β -methyl- α,β -unsaturated Ester **10b** with Organocyanocuprates

entry	reagent ^{a,b}	solvent	product ratio ^c		yield ^d (%)
			11c:11d:11a+11b:12		
1	<i>e</i>	THF	27:69	- <i>g</i> :4	65
2	<i>f</i>	THF	41:54	- <i>g</i> :5	87
3	<i>e</i>	Et ₂ O			<i>h</i>
4	<i>f</i>	Et ₂ O	7:41:26:26		68

^a Reactions in entries 1, 3, and 4 were carried out at -78°C , 0.5 h then 0°C , 3 h with 4 molar equiv of reagent. ^b Reaction in entry 2 was carried out at -78°C , 0.5 h with 4 molar equiv of reagent. ^c Product ratios were determined by HPLC and ^1H NMR. ^d Combined yield. ^e *i*-PrCu(CN)MgCl \cdot BF₃. ^f *i*-Pr₂Cu(CN)(MgCl)₂ \cdot BF₃. ^g Although we cannot conclusively rule out its presence, we failed to isolate the corresponding product. ^h The starting material was recovered.

product with a considerable amount of a D-Phe- ψ [(*Z*)-CH=CMe]-Gly-type dipeptide isostere **12** (entry 4).

The olefinic geometries of **11a–d** were established by NOE experiment, and the stereochemistry of **11a–d** was determined by circular dichroism in a fashion analogous to the determination of the α -alkyl group configuration in acyclic α -alkyl- β,γ -unsaturated esters.¹³ Whereas **11a** and **11d** had negative Cotton effects around 220 nm, **11b** and **11c** exhibited positive Cotton effects as expected. Thus, the alkylated products **11a–d** were identified as the (*2R,3E*)-, (*2S,3Z*)-, (*2S,3E*)-, (*2R,3Z*)-isomers, respectively. The absolute configuration of **11d** was also confirmed by X-ray analysis.

A plausible reaction mechanism is depicted in Figure 2. The predominant formation of the (*2R,3E*)-isomer **11a** and (*2S,3Z*)-isomer **11b** from **10a** and of (*2S,3E*)-isomer **11c** and (*2R,3Z*)-isomer **11d** from **10b** suggested that the products were obtained via *anti*-S_N2' alkylation as in the case of

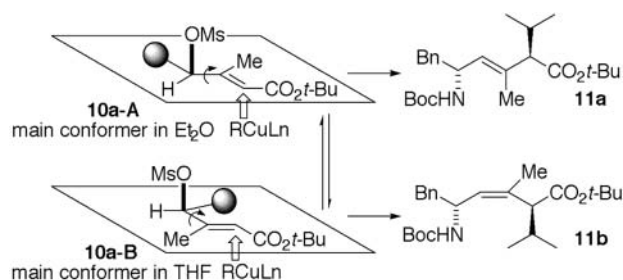


Figure 2. Plausible mechanism for the stereoselective alkylation of *syn*- γ -mesyloxy- β -methyl- α,β -unsaturated ester **10a**.

β -unsubstituted substrates.^{4,11} The products **11a** and **11b** are presumably formed via conformer **10a-A** and conformer **10a-B**, respectively. The ratio of products is assumed to be determined by the population of these conformers in the transition state.¹⁴

In conclusion, we have accomplished the synthesis of ψ [(*E*)-CH=CMe]- and ψ [(*Z*)-CH=CMe]-type dipeptide isosteres via organocopper-mediated alkylation with solvent-dependent geometric selectivity. These isosteres may afford valuable tools for restriction of peptide bonds to *trans*- and *cis*-conformation, respectively. They may be vital for the evaluation of effects of *N*-methylamino acids on conformation of peptides.

Acknowledgment. We thank Dr. Terrence R. Burke, Jr., NCI, NIH, for valuable discussions. This work was supported by Grant-in-Aid for Scientific Research from the Ministry of Education, Culture, Sports, Science, and Technology, Japan, and the Japan Health Science Foundation. S.O. is grateful for Research Fellowships of the JSPS for Young Scientists.

Supporting Information Available: Selected experimental procedures, ^1H NMR spectra for all new compounds, CD spectra of **11a–d**, and crystal structure of **11d**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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(14) Though we cannot rule out the formation of copper- π -allyl complex as a reactive intermediate, it is assumed that the alkylation of the mesylates **10a** and **10b** predominantly proceeded via direct alkylation by organocopper reagents because the alkylation mainly gave *E*- and *Z*-isomers of the *anti*-S_N2' products.